

How do high-performance Li metal batteries perform under low-temperature and high-rate-charging conditions?

Here, we report on high-performance Li metal batteries under low-temperature and high-rate-charging conditions. The high performance is achieved by using a self-assembled monolayer of electrochemically active molecules on current collectors that regulates the nanostructure and composition of the SEI and deposition morphology of Li metal anodes.

How accurate are low-temperature battery models?

In addition to studying the performance of batteries at low temperatures, researchers have also investigated the low-temperature models of batteries. The accuracy of LIB models directly affects battery state estimation, performance prediction, safety warning, and other functions.

What are the advantages of a low-temperature battery?

The prerequisite to support low-temperature operation of batteries is maintaining high ionic conductivity. In contrast to the freezing of OLEs at subzero temperatures, SEs preserve solid state over a wide temperature range without the complete loss of ion-conducting function, which ought to be one of potential advantages.

How do lithium metal batteries perform in a low-temperature environment?

The electrochemical performances of lithium metal batteries are determined by the kinetics of interfacial de-solvation and ion transport, especially at low-temperature environments. Here, a novel electrolyte that easily de-solvated and conducive to interfacial film formation is designed for low-temperature lithium metal batteries.

How bad is a battery at low temperature?

In terms of degradation, the degradation of the battery at low temperature is more serious than at room temperature, and the maximum degradation rate can be 47 times that of room temperature, which increases exponentially as the temperature decreases.

What is a low-temperature battery (LIB)?

They are widely used in different kinds of new-energy vehicles, such as hybrid electric vehicles and battery electric vehicles. However, low-temperature (-20--80 °C) environments hinder the use of LIBs by severely deteriorating their normal performance.

Over the past years, remarkable progress has been achieved at moderate and high temperatures, while the low-temperature operation of all-solid-state batteries emerges as a critical challenge that restricts their wide ...

LiFePO<sub>4</sub> is one of the most widely used cathode materials for lithium-ion batteries, and the low-temperature

performance of LiFePO<sub>4</sub>-based batteries has been widely studied in recent years. Herein, a 3.5 Ah pouch-type full battery was assembled using LiFePO<sub>4</sub> as the cathode and artificial graphite as the anode. For the LiFePO<sub>4</sub>-based cathode, carbon ...

Here, we thoroughly review the state-of-the-arts about battery performance decrease, modeling, and preheating, aiming to drive effective solutions for addressing the low-temperature challenge of LIBs.

Stable operation of rechargeable lithium-based batteries at low temperatures is important for cold-climate applications, but is plagued by dendritic Li plating and unstable solid-electrolyte...

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According to the climate in China (Fig. 1), the regions with the winter temperature below -20 °C account for 38% of the total territory [7] sides, the lowest temperature in high-latitude countries can reach about -40 °C. Therefore, low-temperature LIBs used in civilian field need to withstand temperatures as low as -40 °C (Fig. 1).

Over the past years, remarkable progress has been achieved at moderate and high temperatures, while the low-temperature operation of all-solid-state batteries emerges as a critical challenge that restricts their wide temperature application.

Both low temperature and high temperature that are outside of this region will lead to degradation of performance and irreversible damages, such as lithium plating and thermal runaway. Therefore, understanding the temperature effects and accurate measurement of temperature inside lithium-ion batteries are important for the proper battery management. The ...

Temperature effects on lithium battery performance. Performance at Low Temperatures. In cold temperatures, like below 15 °C (59 °F), lithium batteries experience reduced performance. Chemical reactions within the battery slow down, causing decreased power output. Shorter battery life and diminished capacity result from these conditions. Devices may shut ...

With the rapid development of new-energy vehicles worldwide, lithium-ion batteries (LIBs) are becoming increasingly popular because of their high energy density, long cycle life, and low self-discharge rate. They are ...

Low-temperature operation (-20 °C and below) under high-rate conditions is a critical deficiency for lithium-ion batteries. To achieve size, weight, and power requirements tailored for demanding applications,

novel materials are needed to sustain high performance.

High-performance Li-ion/metal batteries working at a low temperature (i.e.,  $<-20\text{ }^\circ\text{C}$ ) are desired but hindered by the sluggish kinetics associated with  $\text{Li}^+$  transport and charge transfer.

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Especially, the cell has a discharge capacity of  $28.2\text{ mAh g}^{-1}$  after 50 cycles at  $-40\text{ }^\circ\text{C}$  with a coulombic efficiency of  $\sim 99\%$ , which is superior to most low-temperature solid electrolytes. This demonstration of utilizing ZIL as liquid medium for GE may shed light on the development of high-performance low-temperature lithium metal batteries.

Especially under severe conditions of high mass-loading or low-temperature environment, the as-prepared full cell with NH<sub>2</sub>-decorated MOFs exhibits superior electrochemical performance with 90.5% capacity retention for 300 cycles under  $0\text{ }^\circ\text{C}$  and low N/P ratio of 3.3. Even decreasing the temperature down to  $-20\text{ }^\circ\text{C}$ , the capacity-retention of 97% is ...

Extensive research has shown that the electrolyte/electrode composition and microstructure are of fundamental importance to low-temperature performances of LIBs. In this report, we review the recent findings in the role of electrolytes, anodes, and cathodes in the low temperature performances of LIBs.

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